

Sustainable Oxygen Extraction from Lunar Regolith via Molten Salt Electrolysis

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Space Resources Roundtable

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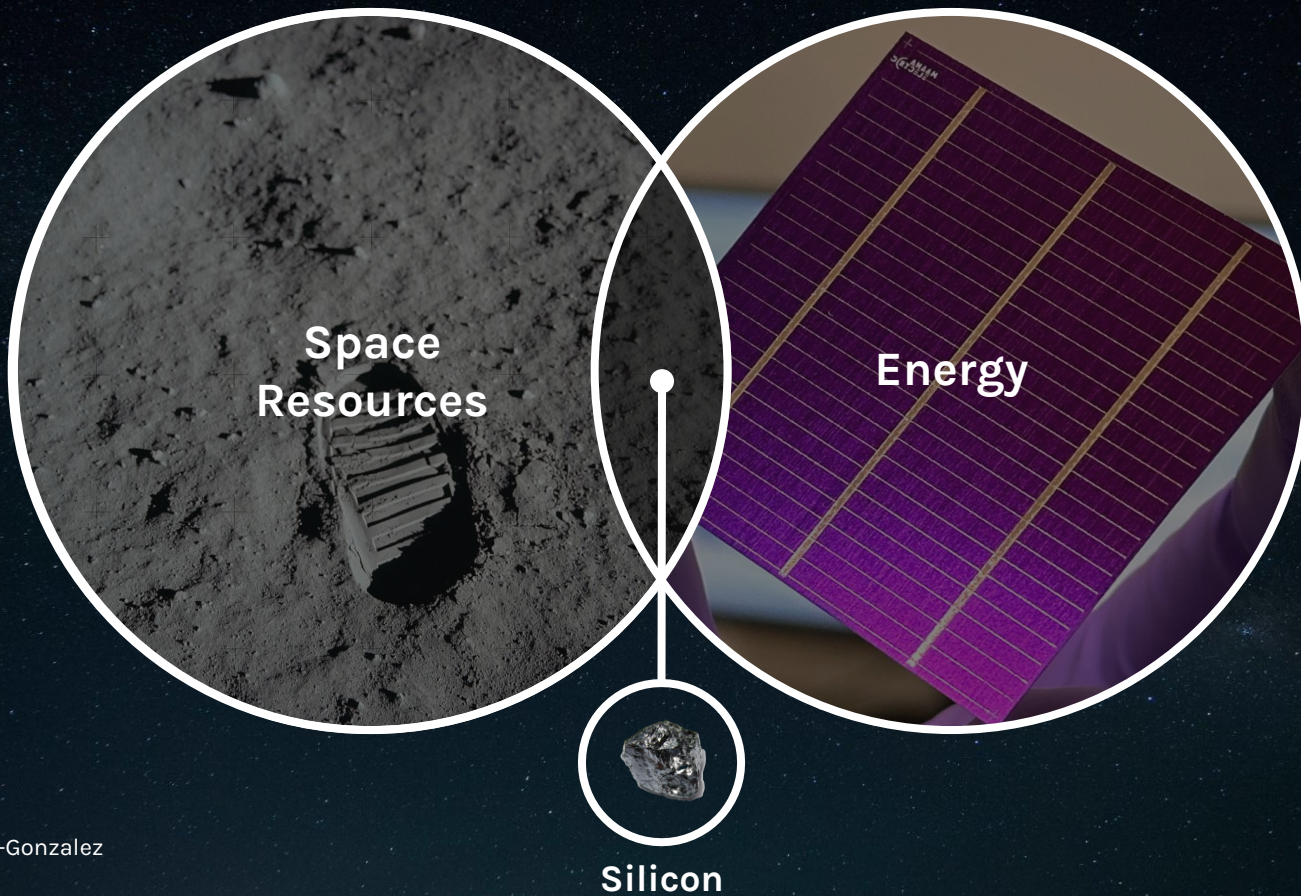
- Founded in 2018.
- Based in Luxembourg, LU 
- 20+ employees
- Specialized in developing planetary surface assets for sustainable space exploration.
- Experienced in full design life cycle and in-house product development (concept, design, manufacturing, and testing), especially TRL 1 to 7.



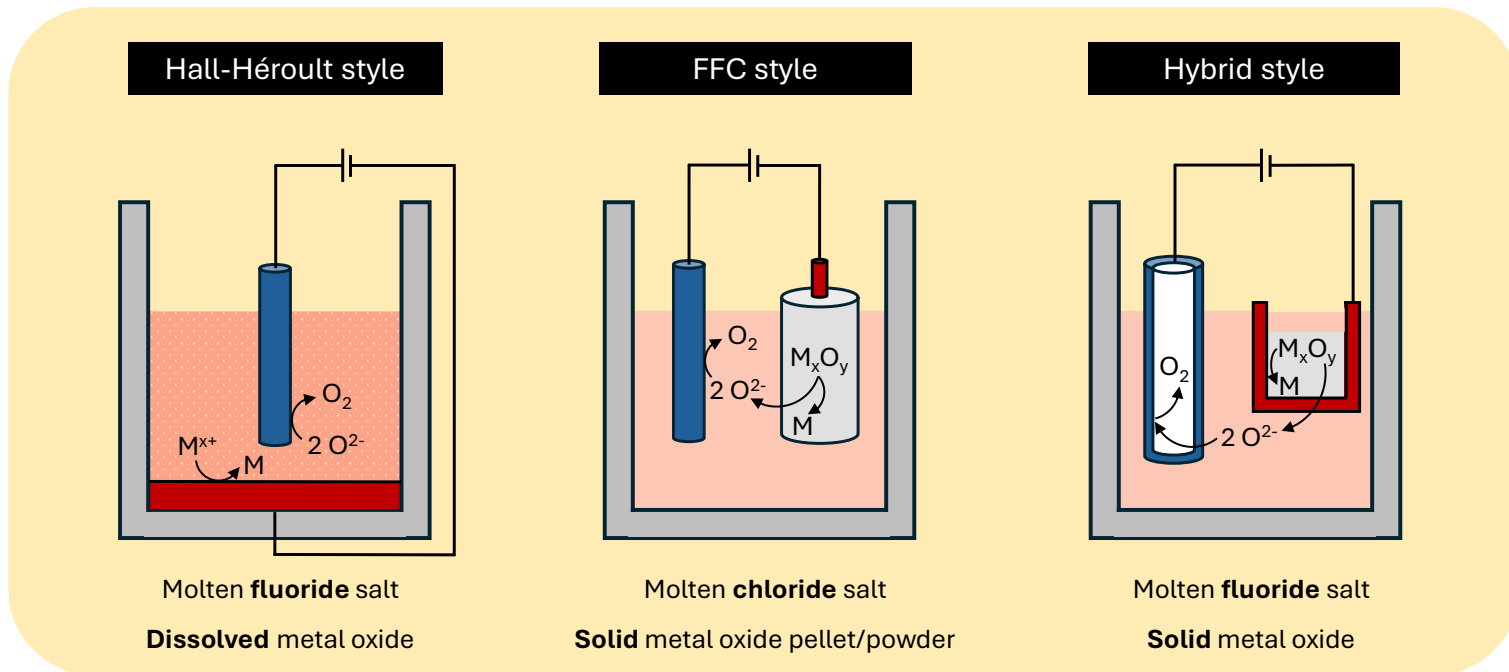
Join my poster on Thursday 5 PM
*Maana Electric's Technologies for Space Resource
Utilization: A European Private-Sector Perspective*

MAANA IN A NUTSHELL

At the intersection of energy and resource utilization

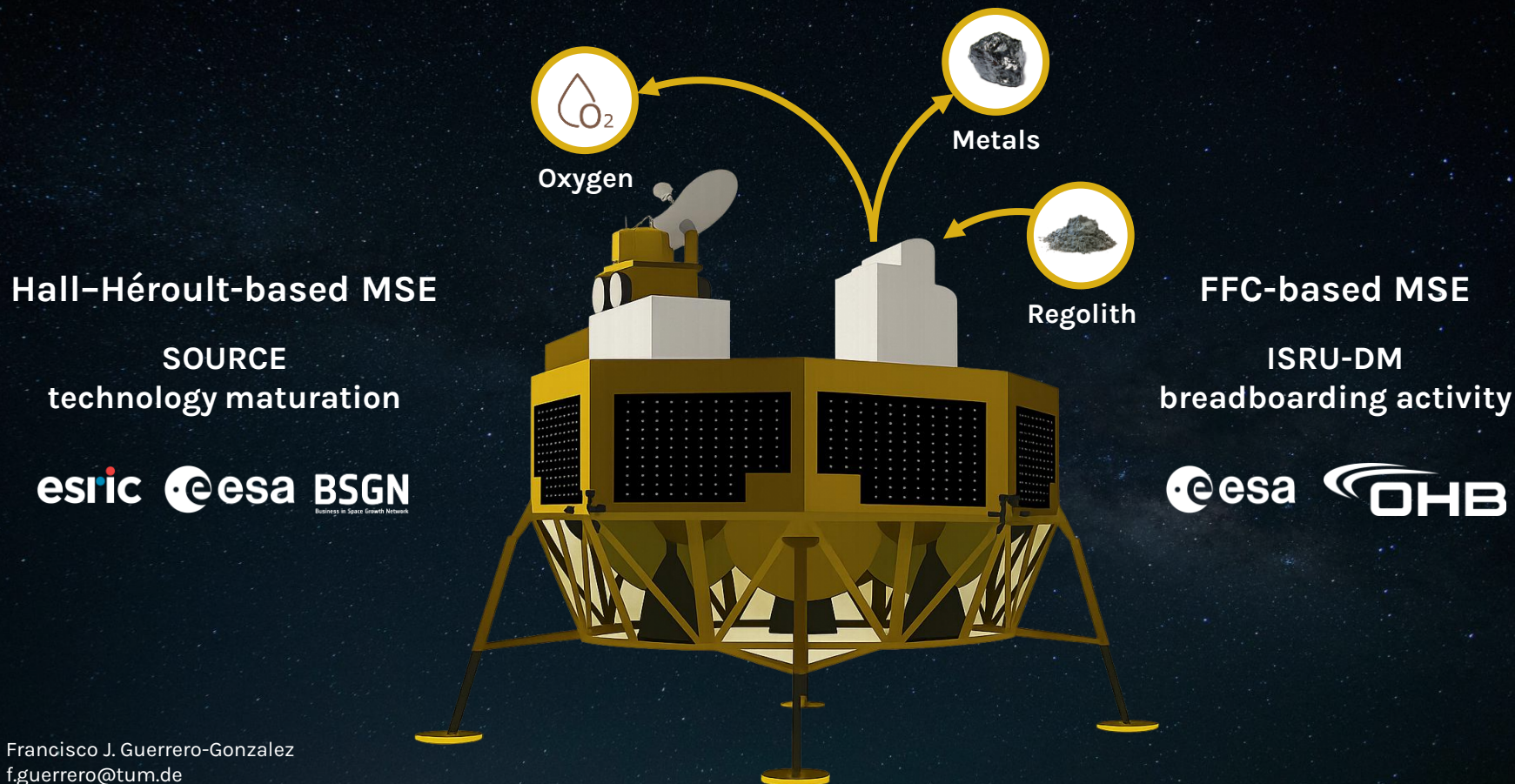


Molten Salt Electrolysis (MSE)



Guerrero-Gonzalez, F.J. et al. (2026). *Electrochemical processing of planetary regolith in molten salts - A review*. **Journal of the Electrochemical Society**, 173(7), 072502. DOI:10.1149/1945-7111/ae5420.

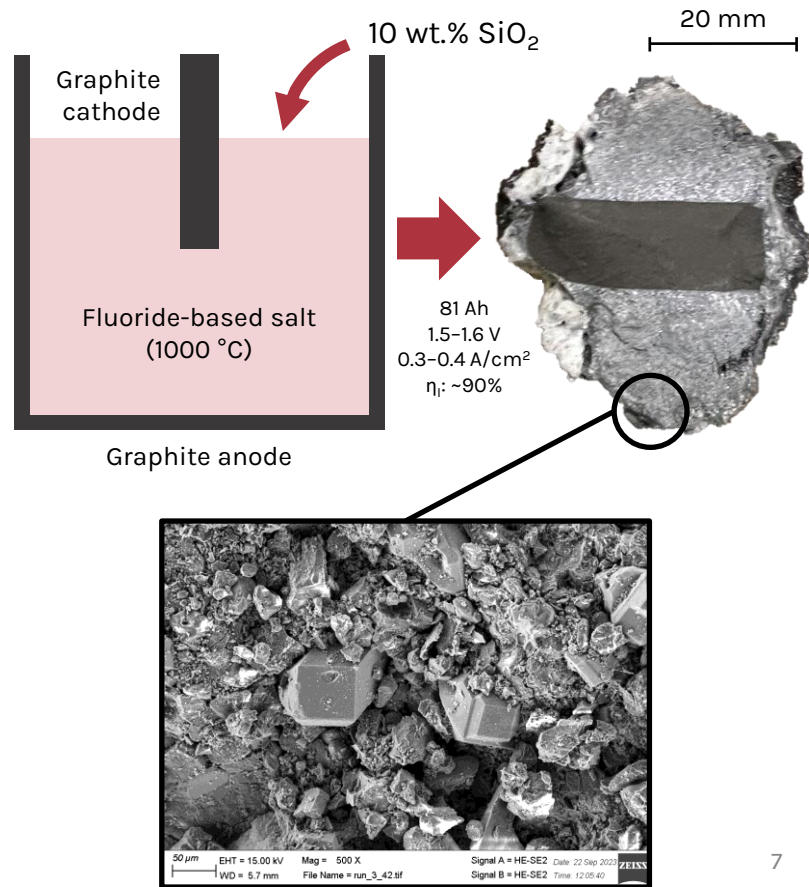
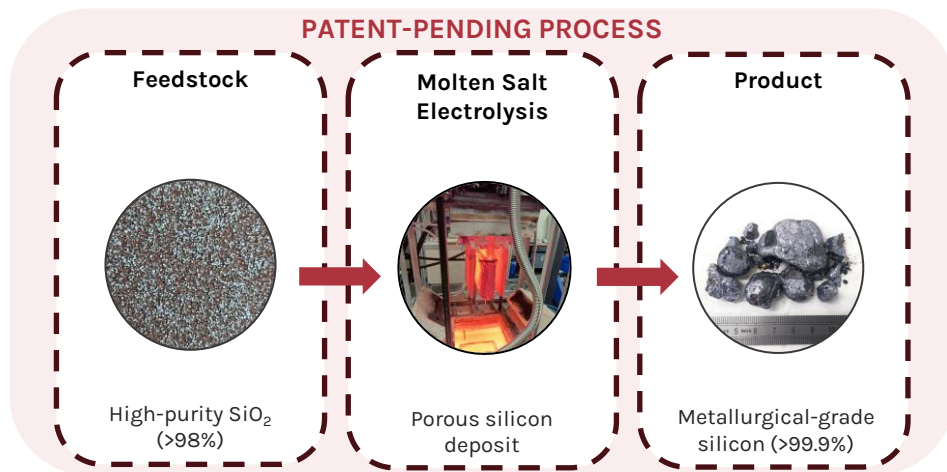
ISRU demonstration missions



Production of Si from low-value feedstock



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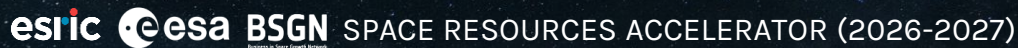


Translating technologies from Earth to space

- ▶ Process maturation for silicon and O₂ from lunar regolith.



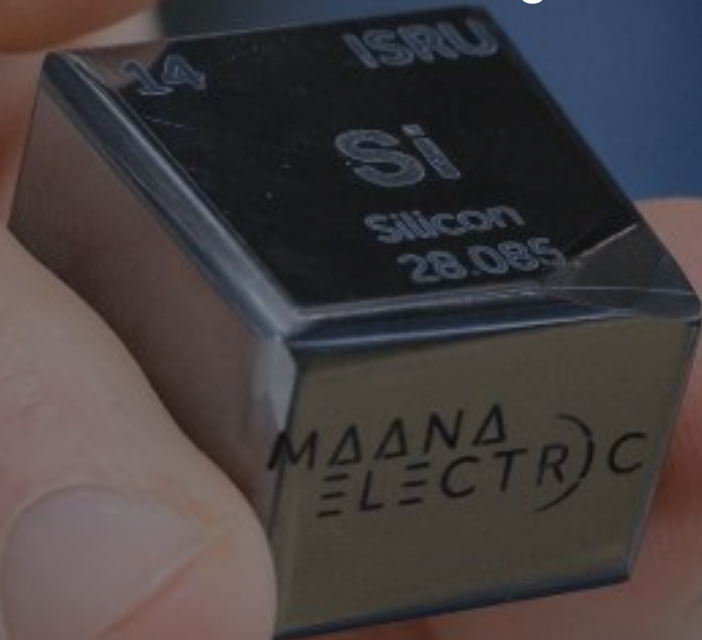
- ▶ Technology demonstration in relevant environment (TRL 5).



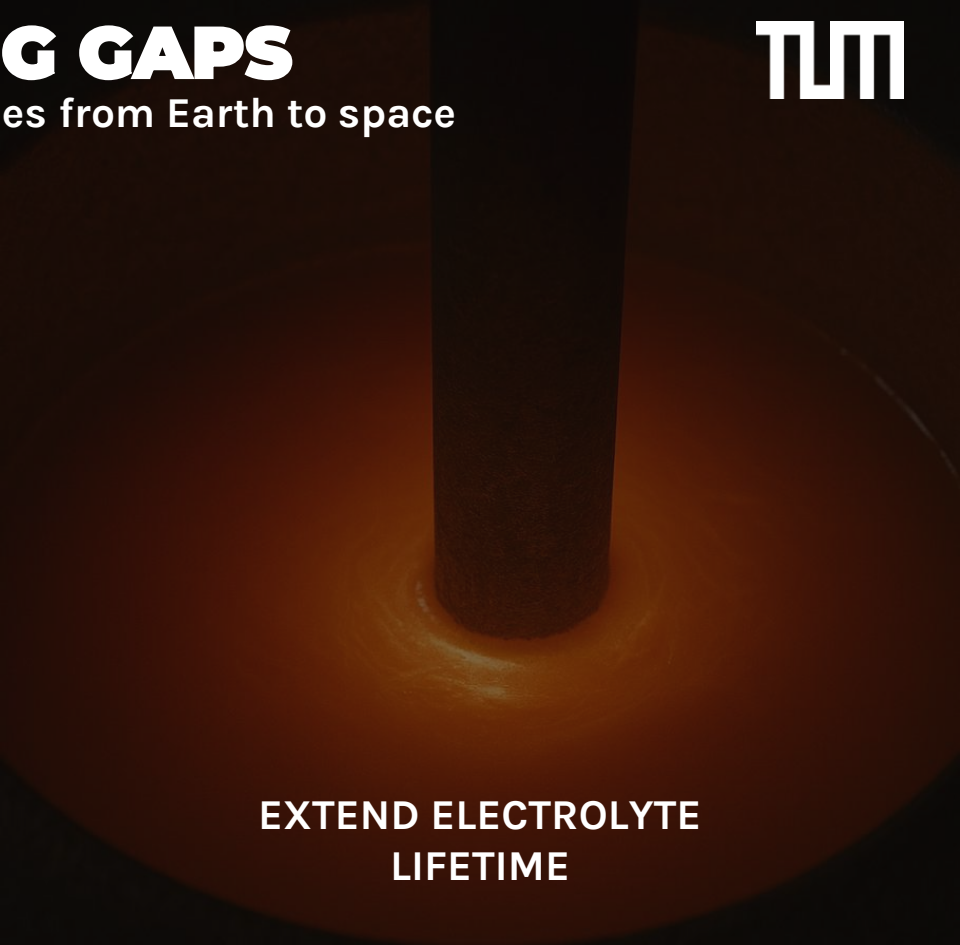
- ▶ Demonstration mission: 50 g of silicon and O₂ on the Moon.
(2029+)

CLOSING GAPS

Translating technologies from Earth to space



IDENTIFY SUITABLE
OXYGEN-EVOLVING ANODES



EXTEND ELECTROLYTE
LIFETIME

Bath poisoning

- Electrolytes shall allow for the reduction of all minerals present in lunar regolith before decomposing themselves.
- Alkaline earth metals (Ca, Mg) are the hardest to reduce.
- CaF_2 could reduce all lunar regolith compounds, preventing bath poisoning.
- CaF_2 melting point is high (1418 °C).

Average elemental composition of Highland regolith (Stoesser et al., 2010).

Mainly from anorthosite →

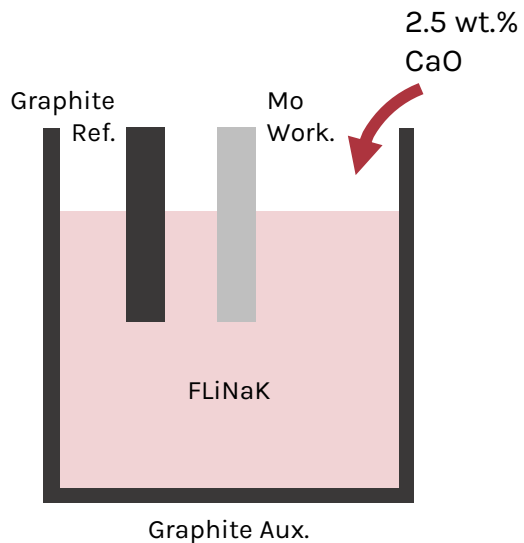
Mainly from pyroxenes & olivines →

	Highland (wt.%)
O	44.6
Si	21.0
Al	13.3
Ca	10.7
Fe	4.9
Mg	4.6
Ti	0.3
Other	0.6

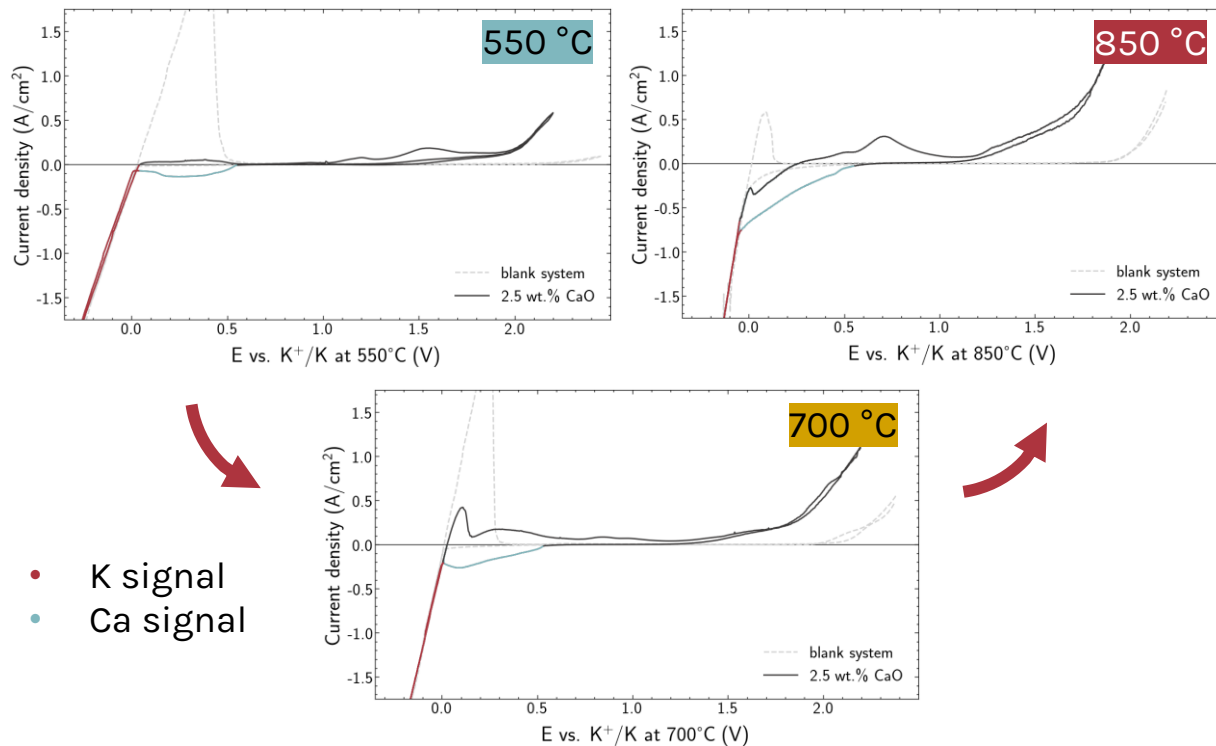
Melting point of fluoride-based salts. Eutectic binary and ternary mixtures.

	Melting point
CaF_2	1418 °C
LiF- CaF_2	770 °C
LiF-NaF	649 °C
LiF-KF	493 °C
FLiNaK	454 °C

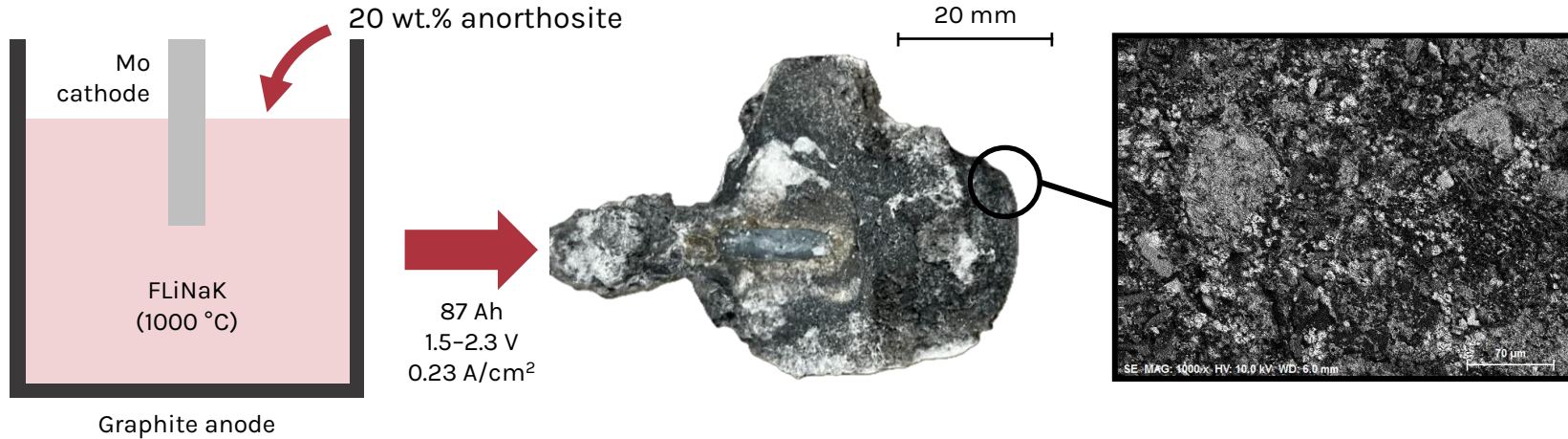
Reduction behavior of Ca^{2+} in FLiNaK



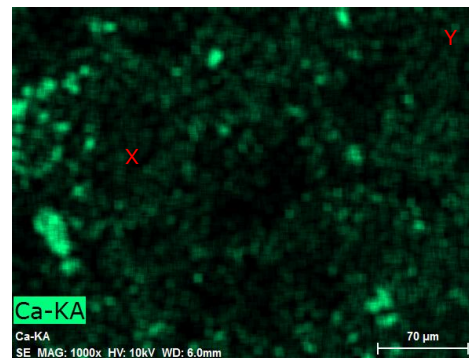
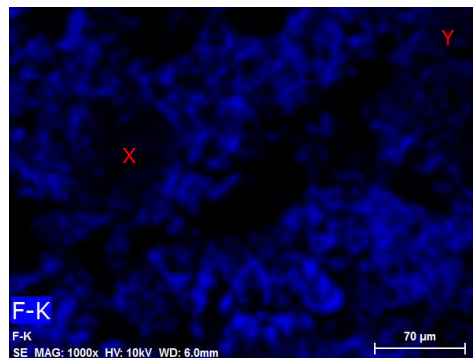
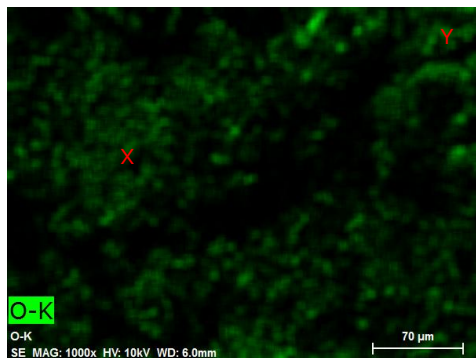
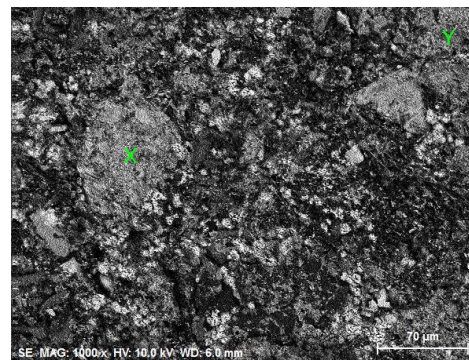
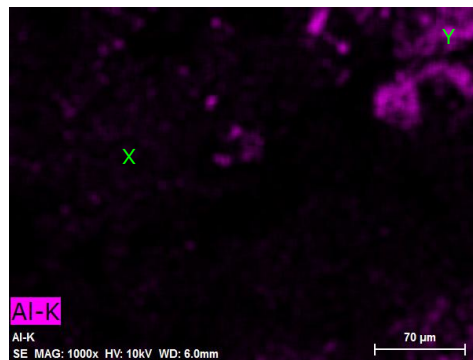
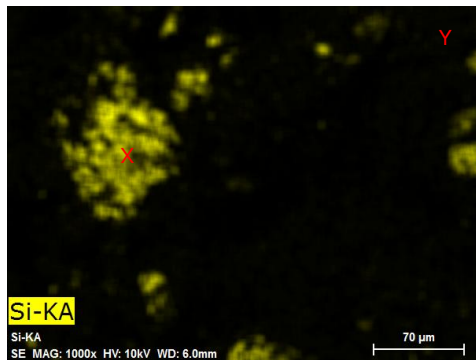
Cyclic voltammograms of FLiNaK-CaO (2.5 wt.%) on Mo electrode at 100 mV/s



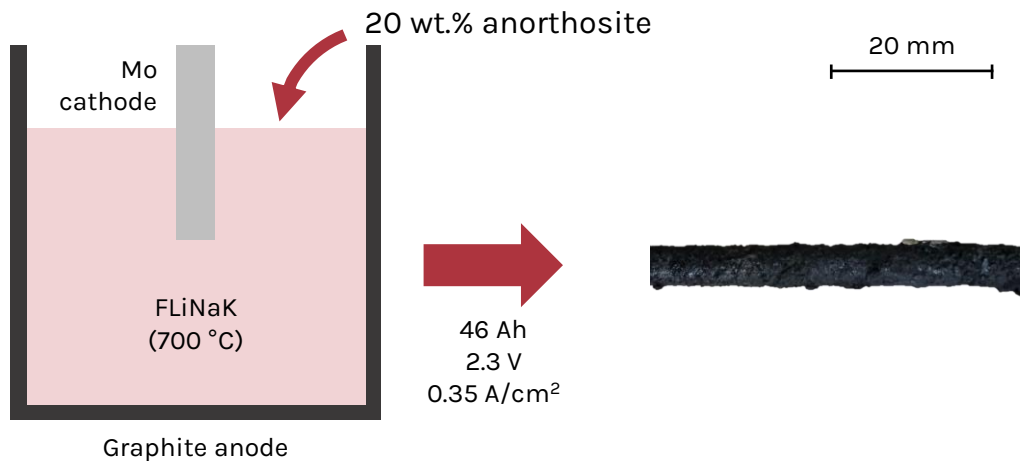
Reduction of lunar anorthosite simulant in FLiNaK



Reduction of lunar anorthosite simulant in FLiNaK



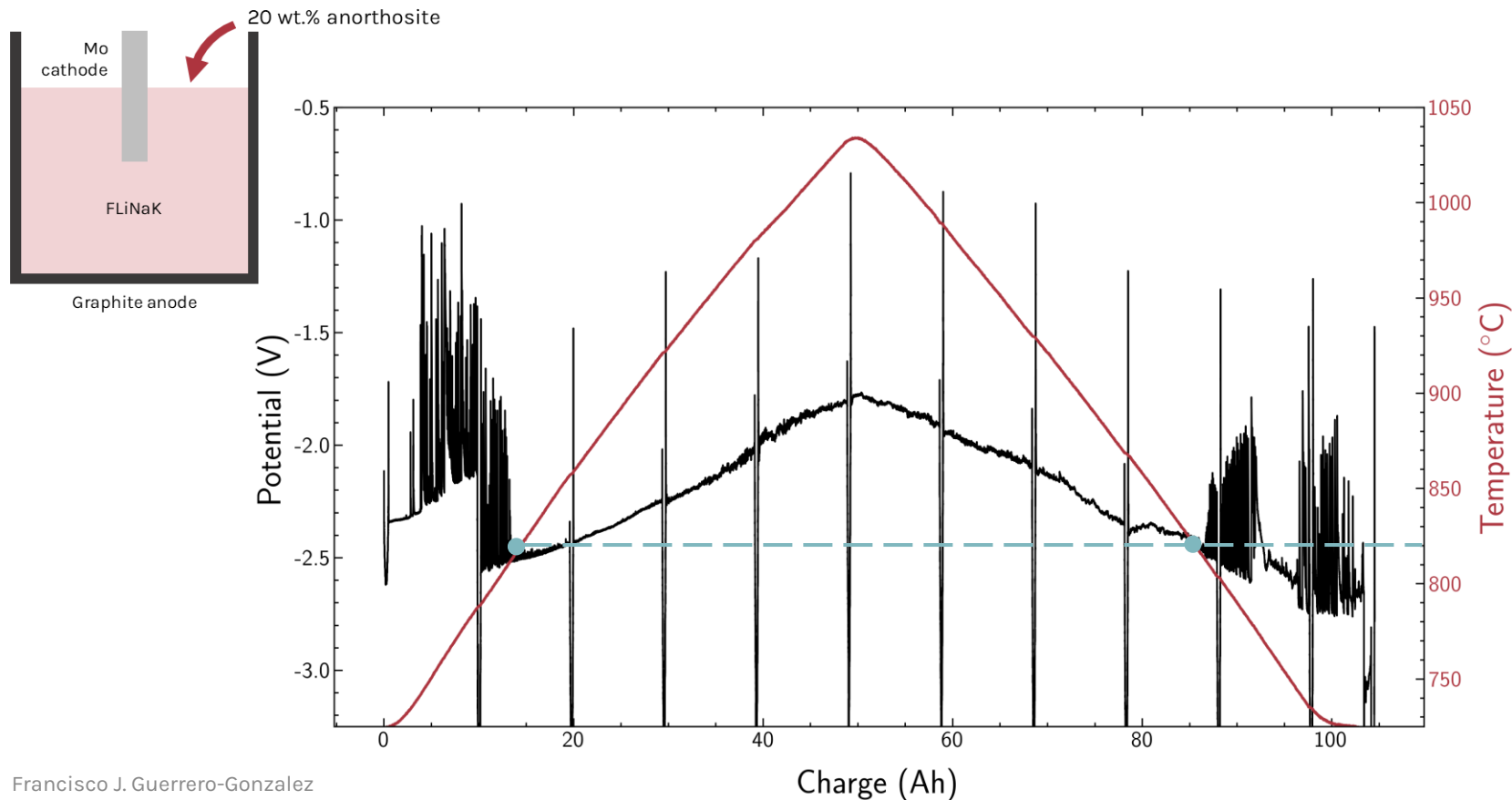
Role of temperature: 700 °C



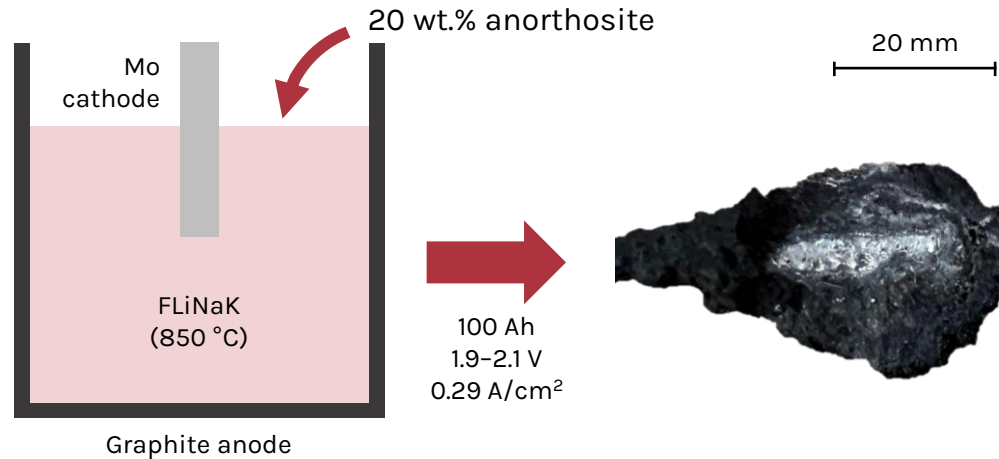
Volatile Si(II) species
Partial Si(IV) → Si(II) reduction



Role of temperature: From 700 °C to 1000 °C and back

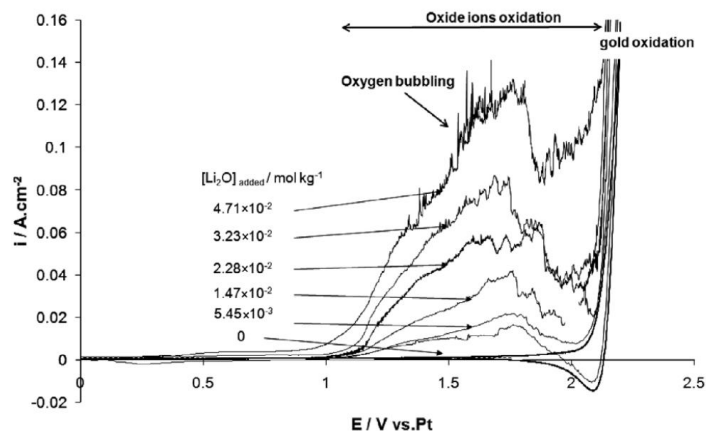


Role of temperature: 850 °C

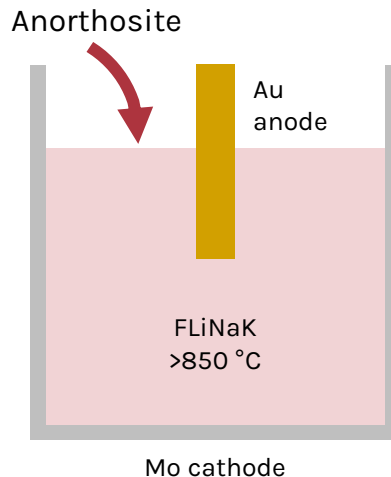


Gold as candidate electrode material

- The principal pathway for anodic chemical corrosion in fluoride melts is *dissolution of oxidized anode material in the melt* (Guerrero-Gonzalez et al. 2026)
- Gold oxides are thermally unstable $>227\text{ }^{\circ}\text{C}$, allowing O_2 evolution
- Gold melting point is low ($1064\text{ }^{\circ}\text{C}$).



Cyclic voltammograms of LiF-NaF-Li₂O on Au electrode at 100 mV/s and 800 °C (Massot et al., 2007)



Sustainable Oxygen Extraction from Lunar Regolith via MSE

- **Calcium can be reduced in FLiNaK** at different temperatures, preventing bath poisoning in the long term.
- **Low-temperature electrolysis** at 850 °C allows reduction of lunar anorthosite simulants into metals. However, higher temperatures result in better products.
- **Selective deposition of silicon** without aluminum co-deposition in FLiNaK has been proven.
- **Current work** is tackling:
 - Oxygen-evolving gold anode testing at >850 °C.
 - Evaluation of alternative candidate materials for increased operational temperature.